



1
00:00:00,790 --> 00:00:07,320

[Music]

2
00:00:12,269 --> 00:00:09,150

[Applause]

3
00:00:15,820 --> 00:00:12,279

thank you I'd like to acknowledge my

4
00:00:17,589 --> 00:00:15,830

colleagues that are contributors to the

5
00:00:22,569 --> 00:00:17,599

work that I will present and I've listed

6
00:00:24,279 --> 00:00:22,579

in here so if we want to understand what

7
00:00:27,249 --> 00:00:24,289

sort of atmospheres to expect around

8
00:00:29,619 --> 00:00:27,259

ocean exoplanets oceans

9
00:00:33,310 --> 00:00:29,629

what sort of biochemistry in their in

10
00:00:34,720 --> 00:00:33,320

their ocean for oceans that have enough

11
00:00:36,400 --> 00:00:34,730

water to have a high-pressure eyes

12
00:00:38,680 --> 00:00:36,410

they're separating the ocean from the

13
00:00:41,560 --> 00:00:38,690

rocky interior we need to introduce a

14

00:00:44,529 --> 00:00:41,570

lot of hydrophobic chemical species

15

00:00:47,380 --> 00:00:44,539

salts which would actually introduce a

16

00:00:50,439 --> 00:00:47,390

lot of crystal structures like clathrate

17

00:00:52,059 --> 00:00:50,449

hydrates filled eyes salty eyes that may

18

00:00:54,819 --> 00:00:52,069

actually fundamentally change our

19

00:00:58,599 --> 00:00:54,829

understanding of how material is being

20

00:01:01,930 --> 00:00:58,609

transported along high pressure high

21

00:01:04,719 --> 00:01:01,940

spangles so I wanted to make a point

22

00:01:06,940 --> 00:01:04,729

here that clathrate hydrates have very

23

00:01:08,890 --> 00:01:06,950

different characteristics than pure

24

00:01:10,840 --> 00:01:08,900

water and we've known this for quite a

25

00:01:14,080 --> 00:01:10,850

while what happens at even higher

26
00:01:17,130 --> 00:01:14,090
pressures beyond 1 GPA well some field

27
00:01:19,120 --> 00:01:17,140
Isis for example that form after

28
00:01:20,830 --> 00:01:19,130
clathrates will basically be doing

29
00:01:22,720 --> 00:01:20,840
dissociate but they dissociate and

30
00:01:25,240 --> 00:01:22,730
restructuring the field ice they could

31
00:01:27,600 --> 00:01:25,250
withstand pressures up to 100 GPA

32
00:01:30,910 --> 00:01:27,610
spanning the entire high-pressure Iseman

33
00:01:32,950 --> 00:01:30,920
so what are the sort of characteristics

34
00:01:34,810 --> 00:01:32,960
do they have so we looked at MD

35
00:01:37,930 --> 00:01:34,820
simulation we did an empty simulation

36
00:01:41,020 --> 00:01:37,940
for a filled ice of methane using the

37
00:01:43,780 --> 00:01:41,030
software CP 2k and we use that empty

38
00:01:45,280 --> 00:01:43,790

simulation to get the vibrational

39

00:01:46,900 --> 00:01:45,290

spectrum of the crystal structure of

40

00:01:50,050 --> 00:01:46,910

filled ice and I just wanted to make a

41

00:01:53,230 --> 00:01:50,060

point these are the results we got for

42

00:01:56,800 --> 00:01:53,240

the heat capacity so this would be the

43

00:01:58,900 --> 00:01:56,810

heat capacity for methane filled ice

44

00:02:01,390 --> 00:01:58,910

compared with heat capacities for pure

45

00:02:03,970 --> 00:02:01,400

water we see that it almost sits on the

46

00:02:06,280 --> 00:02:03,980

Dulong petit value which would suggest

47

00:02:08,800 --> 00:02:06,290

that filled ices have a very low Debye

48

00:02:11,740 --> 00:02:08,810

temperature and a sort of this weird

49

00:02:14,100 --> 00:02:11,750

behavior of clathrate seems to propagate

50

00:02:16,170 --> 00:02:14,110

also into the field ices in

51
00:02:18,660 --> 00:02:16,180
need a lot more work done in order to

52
00:02:21,180 --> 00:02:18,670
understand how they influenced the

53
00:02:23,460 --> 00:02:21,190
interior dynamics and and mass transport

54
00:02:27,360 --> 00:02:23,470
of ocean exoplanets when you have a

55
00:02:29,220 --> 00:02:27,370
high-pressure ice mantle so if these

56
00:02:31,950 --> 00:02:29,230
crystal structures that I mentioned

57
00:02:34,860 --> 00:02:31,960
actually exist in the interior a key

58
00:02:38,730 --> 00:02:34,870
question is what sort of solubility do

59
00:02:40,380 --> 00:02:38,740
they impose on the overlying ocean it's

60
00:02:42,720 --> 00:02:40,390
very important because it will tell us

61
00:02:45,210 --> 00:02:42,730
what sort of chemistry to expect in the

62
00:02:49,710 --> 00:02:45,220
ocean what sort of atmosphere would form

63
00:02:53,940 --> 00:02:49,720

around such planets and actually these

64

00:02:57,660 --> 00:02:53,950

solubilities are not very well known so

65

00:02:59,190 --> 00:02:57,670

we took advantage of data that is known

66

00:03:03,180 --> 00:02:59,200

from the chemical literature on

67

00:03:06,270 --> 00:03:03,190

hydration shell energies and we use the

68

00:03:08,850 --> 00:03:06,280

Kramer's theory for Brownian particles

69

00:03:12,210 --> 00:03:08,860

to relate the solubility of CO_2 in

70

00:03:14,729 --> 00:03:12,220

liquid to the solubility of CO_2 in class

71

00:03:17,070 --> 00:03:14,739

rates and yeah I forgot to mention that

72

00:03:20,370 --> 00:03:17,080

what I want to concentrate on a very

73

00:03:22,979 --> 00:03:20,380

specific problem sorry where you have an

74

00:03:28,340 --> 00:03:22,989

ocean where basically the seafloor of

75

00:03:31,710 --> 00:03:28,350

that ocean is clathrates rich in CO_2 so

76

00:03:34,140 --> 00:03:31,720

we built a theory that could actually

77

00:03:35,940 --> 00:03:34,150

tell us what the solubility of CO₂ in

78

00:03:36,840 --> 00:03:35,950

the liquid would be when in equilibrium

79

00:03:39,650 --> 00:03:36,850

with clathrates

80

00:03:42,060 --> 00:03:39,660

and when we compare this theory with

81

00:03:45,120 --> 00:03:42,070

experiments at low pressure up to a few

82

00:03:47,759 --> 00:03:45,130

hundred bars or model is the blue curve

83

00:03:50,340 --> 00:03:47,769

it goes through experimental data pretty

84

00:03:52,650 --> 00:03:50,350

well and we see a very distinct behavior

85

00:03:54,690 --> 00:03:52,660

where the solubility actually increases

86

00:03:56,660 --> 00:03:54,700

with temperature very different when

87

00:03:59,310 --> 00:03:56,670

class rates are not part of the problem

88

00:04:02,130 --> 00:03:59,320

what happens at higher pressure what

89

00:04:04,020 --> 00:04:02,140

happens is the actual seafloor that's

90

00:04:06,270 --> 00:04:04,030

that's a really big problem because we

91

00:04:11,310 --> 00:04:06,280

don't have direct information for the

92

00:04:14,280 --> 00:04:11,320

search for the solubility on the on the

93

00:04:17,780 --> 00:04:14,290

seafloor at a pressure of 1 GPA we can

94

00:04:20,699 --> 00:04:17,790

only infer that data from melting

95

00:04:23,310 --> 00:04:20,709

melting curve depressions so we took

96

00:04:26,820 --> 00:04:23,320

data from Ballinger in tallinn 2013 for

97

00:04:27,960 --> 00:04:26,830

system saturated in co2 and try to infer

98

00:04:30,090 --> 00:04:27,970

what sort of soluble

99

00:04:31,770 --> 00:04:30,100

you would have wedging equilibrium with

100

00:04:34,760 --> 00:04:31,780

clathrates and it's a very delicate

101
00:04:37,740 --> 00:04:34,770
problem if you assume an ideal sort of

102
00:04:40,800 --> 00:04:37,750
solution you get this green shaded area

103
00:04:42,630 --> 00:04:40,810
if you don't assume ID ality and you

104
00:04:44,760 --> 00:04:42,640
consider the fact that it's and you can

105
00:04:47,460 --> 00:04:44,770
correct for activity coefficients you

106
00:04:49,380 --> 00:04:47,470
get this red shaded area and the model I

107
00:04:51,740 --> 00:04:49,390
showed before with Kramer's theory is

108
00:04:54,510 --> 00:04:51,750
this black curve and it sits very nicely

109
00:04:58,620 --> 00:04:54,520
on the continuation of this inferred

110
00:05:00,450 --> 00:04:58,630
data so what we generally know from this

111
00:05:02,370 --> 00:05:00,460
modeling we only use this model to make

112
00:05:04,020 --> 00:05:02,380
interpolations it's very very important

113
00:05:06,000 --> 00:05:04,030

these models are not good to make

114

00:05:08,220 --> 00:05:06,010

extrapolations the experimental data is

115

00:05:10,320 --> 00:05:08,230

very very important but what we

116

00:05:12,510 --> 00:05:10,330

generally know when you are in equity

117

00:05:15,570 --> 00:05:12,520

room with class rates the solubility is

118

00:05:17,580 --> 00:05:15,580

not a strong function of pressure unlike

119

00:05:20,370 --> 00:05:17,590

when you use the Henry's law for example

120

00:05:22,080 --> 00:05:20,380

and it is reversed with temperature

121

00:05:24,030 --> 00:05:22,090

increases when the temperature increases

122

00:05:29,190 --> 00:05:24,040

this is what clathrates do when they're

123

00:05:30,570 --> 00:05:29,200

in the system so that's a very important

124

00:05:33,960 --> 00:05:30,580

thing to know the song the ability

125

00:05:37,710 --> 00:05:33,970

because if you want to consider the pH

126

00:05:41,400 --> 00:05:37,720

of an ocean you you basically want to

127

00:05:43,230 --> 00:05:41,410

solve this set of equations and you have

128

00:05:45,540 --> 00:05:43,240

five equations you need an you need a

129

00:05:47,880 --> 00:05:45,550

variable to constrain the system because

130

00:05:50,850 --> 00:05:47,890

you have six variables and for these

131

00:05:52,740 --> 00:05:50,860

particular planets if you have a sea

132

00:05:55,800 --> 00:05:52,750

floor that's composed of CO_2 it's an

133

00:05:58,590 --> 00:05:55,810

open system so if it becomes unsaturated

134

00:05:59,969 --> 00:05:58,600

with CO_2 CO_2 will enter the ocean

135

00:06:01,969 --> 00:05:59,979

because the clathrates will not be

136

00:06:05,219 --> 00:06:01,979

stable they will keep the ocean

137

00:06:07,350 --> 00:06:05,229

saturated so your solubility is

138

00:06:10,230 --> 00:06:07,360

basically what's governing the pH of the

139

00:06:13,200 --> 00:06:10,240

ocean now here they're charged neutral

140

00:06:15,719 --> 00:06:13,210

any equation that I've written seems

141

00:06:18,240 --> 00:06:15,729

extremely simple it's not true for Earth

142

00:06:20,610 --> 00:06:18,250

where you have a lot of interesting ions

143

00:06:22,320 --> 00:06:20,620

dissolved in the ocean and I want to

144

00:06:25,140 --> 00:06:22,330

touch on

145

00:06:31,380 --> 00:06:25,150

basically explain why we think it may

146

00:06:34,110 --> 00:06:31,390

have this simple form so recently there

147

00:06:38,430 --> 00:06:34,120

has been a lot of studies that have been

148

00:06:40,830 --> 00:06:38,440

published on what happens to Bryan's

149

00:06:43,590 --> 00:06:40,840

solutions at very high pressure

150

00:06:45,930 --> 00:06:43,600

and it seems that if the very

151

00:06:47,670 --> 00:06:45,940

high-pressure the end result is very

152

00:06:49,890 --> 00:06:47,680

different than than from low pressure

153

00:06:53,129 --> 00:06:49,900

where the salt is exalt out of the

154

00:06:58,170 --> 00:06:53,139

structure in very high pressure where I

155

00:06:59,900 --> 00:06:58,180

6 and I 7 are concerned ions dissolved

156

00:07:02,219 --> 00:06:59,910

in the liquid actually become

157

00:07:04,379 --> 00:07:02,229

incorporated as interstitials in the

158

00:07:07,230 --> 00:07:04,389

voids of high pressure isa

159

00:07:10,379 --> 00:07:07,240

jurua Tala studied the system Franco

160

00:07:12,870 --> 00:07:10,389

colander other studies and it's more so

161

00:07:17,310 --> 00:07:12,880

it's more correct for i7 than for I 6

162

00:07:21,320 --> 00:07:17,320

though but an interesting another

163

00:07:24,150 --> 00:07:21,330

interesting experimental result is that

164

00:07:26,879 --> 00:07:24,160

and talk about a paper public by Frank

165

00:07:29,360 --> 00:07:26,889

at al 2008 if you take such a high

166

00:07:32,580 --> 00:07:29,370

pressure I swear it has interstitials

167

00:07:34,890 --> 00:07:32,590

within its voids of ions of salt when

168

00:07:37,620 --> 00:07:34,900

you increase the pressure to 500 K it

169

00:07:40,680 --> 00:07:37,630

seems like that the ions are exalt out

170

00:07:42,990 --> 00:07:40,690

of the ice and form pure grains of dense

171

00:07:45,930 --> 00:07:43,000

salt why is that important and how is

172

00:07:47,850 --> 00:07:45,940

that what does that imply for ocean

173

00:07:51,930 --> 00:07:47,860

exoplanets that have a high pressure ice

174

00:07:53,909 --> 00:07:51,940

mantle so we solved for the thermal

175

00:07:56,659 --> 00:07:53,919

profile let's say this is the bottom

176

00:07:59,640 --> 00:07:56,669

this is the sea floor and then you have

177

00:08:01,680 --> 00:07:59,650

high pressure ice this is the thermal

178

00:08:04,830 --> 00:08:01,690

profile in the high pressure Einstein 6

179

00:08:07,650 --> 00:08:04,840

and I seven and this condition that I've

180

00:08:11,400 --> 00:08:07,660

written here is basically saying that if

181

00:08:13,050 --> 00:08:11,410

this condition is true which for these

182

00:08:16,219 --> 00:08:13,060

planets is true even if you have tiny

183

00:08:18,930 --> 00:08:16,229

fractions of rock in the planet then

184

00:08:21,690 --> 00:08:18,940

basically this is a scaling for the

185

00:08:23,580 --> 00:08:21,700

thermal conductivity and the temperature

186

00:08:25,920 --> 00:08:23,590

gradient of the melting of the melting

187

00:08:27,810 --> 00:08:25,930

curve that the bottom of the ocean the

188

00:08:30,029 --> 00:08:27,820

sea floor is actually constrained to the

189

00:08:33,480 --> 00:08:30,039

melting temperature the the melting

190

00:08:35,699 --> 00:08:33,490

curve of high pressure ice but being

191

00:08:37,740 --> 00:08:35,709

constrained to the melting curve of high

192

00:08:41,730 --> 00:08:37,750

pressure ice places are constrained on

193

00:08:43,769 --> 00:08:41,740

your ability to conductively cool and

194

00:08:46,860 --> 00:08:43,779

what this means is that high pressure

195

00:08:49,410 --> 00:08:46,870

ice that is upwelling would at some

196

00:08:52,360 --> 00:08:49,420

point tend to melt and basically

197

00:08:55,150 --> 00:08:52,370

discharge it's hot water into the ocean

198

00:08:57,190 --> 00:08:55,160

but if you're not basically melting your

199

00:08:59,920 --> 00:08:57,200

entire ice mantle and you're in a steady

200

00:09:03,070 --> 00:08:59,930

state then at a lower temperature you

201
00:09:04,570 --> 00:09:03,080
have to rican dents so there so I want

202
00:09:07,030 --> 00:09:04,580
you to have this picture in mind where

203
00:09:09,010 --> 00:09:07,040
you have an ocean that's discharging

204
00:09:10,780 --> 00:09:09,020
that you have high-pressure ice melting

205
00:09:12,730 --> 00:09:10,790
discharging water into the ocean and

206
00:09:18,370 --> 00:09:12,740
then rican dents and creating a new

207
00:09:20,770 --> 00:09:18,380
seafloor also another point that i want

208
00:09:22,470 --> 00:09:20,780
to show that contrary to icy moons

209
00:09:25,480 --> 00:09:22,480
because it's very high pressure

210
00:09:27,100 --> 00:09:25,490
adiabatic temperature profiles differ

211
00:09:28,900 --> 00:09:27,110
from the melting temperatures this is

212
00:09:32,020 --> 00:09:28,910
the melting temperature for I seven

213
00:09:35,320 --> 00:09:32,030

differ they started diverging to

214

00:09:36,910 --> 00:09:35,330

hundreds of hundreds of kelvins so if

215

00:09:39,690 --> 00:09:36,920

you want to create for example a brine

216

00:09:42,040 --> 00:09:39,700

pocket and you want to create localized

217

00:09:45,130 --> 00:09:42,050

mounting you need to overcome a

218

00:09:46,360 --> 00:09:45,140

substantial temperature barrier here so

219

00:09:50,140 --> 00:09:46,370

if you look at is Oh melting

220

00:09:52,690 --> 00:09:50,150

temperatures as a function of pressure

221

00:09:54,430 --> 00:09:52,700

the activity of the solvent which is

222

00:09:56,910 --> 00:09:54,440

water here that you need in order to

223

00:10:00,100 --> 00:09:56,920

create such a huge melting de pressure

224

00:10:01,540 --> 00:10:00,110

melting temperature a depression is you

225

00:10:03,820 --> 00:10:01,550

need the solvent to have an activity of

226

00:10:05,920 --> 00:10:03,830

0.2 we know from experiments that water

227

00:10:08,710 --> 00:10:05,930

does not reach these values somewhere

228

00:10:11,350 --> 00:10:08,720

around 0.7 so you cannot create brunt of

229

00:10:12,670 --> 00:10:11,360

pockets of brine so I I totally for a

230

00:10:15,910 --> 00:10:12,680

lot of things and I want to wrap it up

231

00:10:18,880 --> 00:10:15,920

in a in this sort of cartoon on an

232

00:10:20,920 --> 00:10:18,890

illustration that explains basically

233

00:10:23,050 --> 00:10:20,930

what these experimental data and the

234

00:10:25,030 --> 00:10:23,060

thermal profile basically shows us it

235

00:10:28,420 --> 00:10:25,040

shows us that that if high pressure ice

236

00:10:30,970 --> 00:10:28,430

up welds it melts into the ocean forms a

237

00:10:33,490 --> 00:10:30,980

new ocean floor but the new ocean floor

238

00:10:35,800 --> 00:10:33,500

that forms it actually has a high

239

00:10:38,380 --> 00:10:35,810

fractionation factor of ions dissolved

240

00:10:40,480 --> 00:10:38,390

in the ocean with a new ocean floor more

241

00:10:43,780 --> 00:10:40,490

so far it than for I 6 now stroll

242

00:10:46,090 --> 00:10:43,790

results for to in terms of what it means

243

00:10:48,970 --> 00:10:46,100

for the salinity of the ocean but if

244

00:10:51,550 --> 00:10:48,980

this part remixes into the mantle then

245

00:10:53,290 --> 00:10:51,560

when reaching the 500k threshold which

246

00:10:57,360 --> 00:10:53,300

we find that you indeed reach in the

247

00:11:00,550 --> 00:10:57,370

interior these ions that are

248

00:11:02,200 --> 00:11:00,560

interstitials in the ice actually start

249

00:11:04,510 --> 00:11:02,210

to dissolve out of the high pressure

250

00:11:05,480 --> 00:11:04,520

high structure and form grains of salt

251
00:11:08,929 --> 00:11:05,490
that sediment

252
00:11:11,090 --> 00:11:08,939
onto the rocky Court so on the one hand

253
00:11:13,040 --> 00:11:11,100
you have relatively pure water coming

254
00:11:15,499 --> 00:11:13,050
out and you have salt coming in so

255
00:11:18,980 --> 00:11:15,509
there's a pump that may actually start

256
00:11:21,739 --> 00:11:18,990
to desalinate the ocean over time and we

257
00:11:24,109 --> 00:11:21,749
calculated the time scale for this

258
00:11:27,290 --> 00:11:24,119
desalination process it depends on the

259
00:11:29,600 --> 00:11:27,300
fractionation factor of ions between

260
00:11:34,369 --> 00:11:29,610
liquid water and high-pressure ice and

261
00:11:37,730 --> 00:11:34,379
the and the melting rate and if the sea

262
00:11:38,210 --> 00:11:37,740
floor is if the timescale is very very

263
00:11:40,850 --> 00:11:38,220

short

264

00:11:43,489 --> 00:11:40,860

it basically desalinate sea ocean very

265

00:11:45,769 --> 00:11:43,499

rapidly if it's size six which i

266

00:11:48,049 --> 00:11:45,779

think is more reasonable if something

267

00:11:50,150 --> 00:11:48,059

like I six then the time scale becomes

268

00:11:52,309 --> 00:11:50,160

very large but even this large time

269

00:11:53,799 --> 00:11:52,319

scale would suggest that you don't end

270

00:11:56,600 --> 00:11:53,809

up with more than milli molar

271

00:11:58,160 --> 00:11:56,610

concentrations of salt in the ocean so

272

00:12:02,119 --> 00:11:58,170

these oceans could be very rich in

273

00:12:06,460 --> 00:12:02,129

volatiles very poor insults which would

274

00:12:09,410 --> 00:12:06,470

say that the way that we've written our

275

00:12:12,980 --> 00:12:09,420

speciation of the eight of the carbonate

276

00:12:16,220 --> 00:12:12,990

system properly and what this would

277

00:12:19,189 --> 00:12:16,230

suggest is this pH profile in such

278

00:12:21,710 --> 00:12:19,199

oceans so the pH may go from as high as

279

00:12:24,019 --> 00:12:21,720

3.5 at the surface all the way to around

280

00:12:28,730 --> 00:12:24,029

2 at the sea floor so it's a highly

281

00:12:33,319 --> 00:12:28,740

acidic environment and if we plot this

282

00:12:36,340 --> 00:12:33,329

pH temperature die BH temperature that

283

00:12:40,069 --> 00:12:36,350

we have studied on a poly extrema file

284

00:12:42,730 --> 00:12:40,079

plot found that we've taken from kapitza

285

00:12:45,379 --> 00:12:42,740

all we see that it's a pretty barren

286

00:12:47,150 --> 00:12:45,389

region so it could be that it's a rare

287

00:12:48,949 --> 00:12:47,160

environment on earth and that's the

288

00:12:52,129 --> 00:12:48,959

reason why there isn't a lot of life

289

00:12:54,710 --> 00:12:52,139

here it could be substantially a harsh

290

00:12:59,629 --> 00:12:54,720

environment for life but in addition we

291

00:13:02,329 --> 00:12:59,639

have it's a poly it basically is an

292

00:13:04,850 --> 00:13:02,339

environment that is has many stressors

293

00:13:08,090 --> 00:13:04,860

it's not just the agent temperature it

294

00:13:10,670 --> 00:13:08,100

has to combat low salinity as I've just

295

00:13:14,150 --> 00:13:10,680

explained and

296

00:13:17,420 --> 00:13:14,160

way that you may do that is by

297

00:13:20,780 --> 00:13:17,430

introducing an ice camp and a reason

298

00:13:22,880 --> 00:13:20,790

that that may help with some of the

299

00:13:25,970 --> 00:13:22,890

issues of the environment being highly

300

00:13:28,310 --> 00:13:25,980

diluted with salts is with the with is

301

00:13:30,560 --> 00:13:28,320

with the fact that you tactic freezing

302

00:13:33,970 --> 00:13:30,570

within the pores of ice may actually

303

00:13:36,110 --> 00:13:33,980

create niches where you may reach higher

304

00:13:40,880 --> 00:13:36,120

concentrations that may be more

305

00:13:43,850 --> 00:13:40,890

interesting for combating hydrolysis and

306

00:13:56,150 --> 00:13:43,860

creating polymerization and I leave you

307

00:14:04,850 --> 00:13:56,160

with these exclusions a few minutes for

308

00:14:10,079 --> 00:14:08,610

hey as Steve ants again JPL I guess I

309

00:14:12,930 --> 00:14:10,089

didn't understand the mechanism by which

310

00:14:14,400 --> 00:14:12,940

I six exports salt to the bottom it

311

00:14:16,769 --> 00:14:14,410

makes sense to me based on what you said

312

00:14:19,100 --> 00:14:16,779

that I seven would do that the

313

00:14:25,380 --> 00:14:19,110

difference would be inefficiency so

314

00:14:27,360 --> 00:14:25,390

basically the the major point would be

315

00:14:29,280 --> 00:14:27,370

what is the fractionation because this

316

00:14:31,920 --> 00:14:29,290

this actually doesn't depend on I 6 or

317

00:14:33,750 --> 00:14:31,930

i7 right in terms of energy balance

318

00:14:36,360 --> 00:14:33,760

you're gonna have to melt and you're

319

00:14:39,150 --> 00:14:36,370

gonna have to reform a newly a new ice

320

00:14:42,660 --> 00:14:39,160

ocean floor that will mix the point is

321

00:14:44,880 --> 00:14:42,670

how much I salt you actually end up

322

00:14:46,800 --> 00:14:44,890

within this region so the difference is

323

00:14:48,900 --> 00:14:46,810

what is the fractionation factor of

324

00:14:50,850 --> 00:14:48,910

soils between the ocean and the

325

00:14:53,190 --> 00:14:50,860

high-pressure ice that would change

326

00:14:55,139 --> 00:14:53,200

between I six and seven for I seven it

327

00:14:57,360 --> 00:14:55,149

would be very efficient because it has

328

00:14:59,820 --> 00:14:57,370

larger voids for I six it would be

329

00:15:04,139 --> 00:14:59,830

somewhat less it's less efficient but so

330

00:15:07,019 --> 00:15:04,149

that so that's why yeah if I if I apply

331

00:15:09,269 --> 00:15:07,029

to this diagram with uh at Audi it's an

332

00:15:10,350 --> 00:15:09,279

exponential function kind of where i-70

333

00:15:12,090 --> 00:15:10,360

would end up with nothing

334

00:15:14,280 --> 00:15:12,100

okay so I six just has a lower carrying

335

00:15:17,100 --> 00:15:14,290

capacity yeah exact scale exactly and

336

00:15:20,090 --> 00:15:17,110

the key thing about the ocean worlds or

337

00:15:23,940 --> 00:15:20,100

sorry water planet sources ocean moons

338

00:15:25,530 --> 00:15:23,950

is that it pressures and in the water

339

00:15:27,600 --> 00:15:25,540

worlds are high enough that you reach

340

00:15:30,600 --> 00:15:27,610

this dissolution point at the bottom of

341

00:15:34,079 --> 00:15:30,610

the ice yeah I mean there's several I

342

00:15:35,910 --> 00:15:34,089

mean first of all on this arm so the

343

00:15:38,550 --> 00:15:35,920

question so for this pump to work you

344

00:15:42,030 --> 00:15:38,560

have to bring in relatively pure water

345

00:15:45,480 --> 00:15:42,040

and exit and and and you know and and

346

00:15:47,970 --> 00:15:45,490

and and basically sediment the ice right

347

00:15:50,639 --> 00:15:47,980

if you bring the salt back up then you

348

00:15:54,060 --> 00:15:50,649

don't have a pump so for ocean world's

349

00:15:55,870 --> 00:15:54,070

first of all you have a relatively at

350

00:15:58,540 --> 00:15:55,880

you saw the adiabat in the Mel

351

00:16:00,910 --> 00:15:58,550

they really diverge so for icing moons

352

00:16:03,960 --> 00:16:00,920

that temperature is actually because of

353

00:16:06,730 --> 00:16:03,970

the low pressure is not that that

354

00:16:09,100 --> 00:16:06,740

substantial it's a few tens of K you can

355

00:16:11,260 --> 00:16:09,110

easily overcome that and create brined

356

00:16:14,260 --> 00:16:11,270

pockets that can certainly migrate up

357

00:16:17,170 --> 00:16:14,270

certainly if you have like some models

358

00:16:21,370 --> 00:16:17,180

for typing a mixed ice rock layer it

359

00:16:23,550 --> 00:16:21,380

won't be a problem to basically opt well

360

00:16:26,320 --> 00:16:23,560

a brine pocket in this sort of situation

361

00:16:29,890 --> 00:16:26,330

it's very hard to do that you can't

362

00:16:33,310 --> 00:16:29,900

overcome a temperature difference of

363

00:16:38,950 --> 00:16:33,320

hundreds of K due to the high pressure

364

00:16:41,410 --> 00:16:38,960

and and again because beyond 500 K you

365

00:16:44,410 --> 00:16:41,420

don't get the ions as interstitials in

366

00:16:46,510 --> 00:16:44,420

the high pressure ice then you can't get

367

00:16:48,700 --> 00:16:46,520

the ions back as interstitials and you

368

00:16:51,490 --> 00:16:48,710

can't get brine pockets up there as well

369

00:16:53,410 --> 00:16:51,500

so the melting would probably happen

370

00:16:55,120 --> 00:16:53,420

somewhere here where the temperature

371

00:16:57,280 --> 00:16:55,130

difference between the adiabatic and the

372

00:16:59,140 --> 00:16:57,290

melting is not that hot so that's why I

373

00:17:02,740 --> 00:16:59,150

work here but it's not there work for

374

00:17:04,240 --> 00:17:02,750

icy moons is consistent with some recent

375

00:17:07,000 --> 00:17:04,250

literature on the two-phase convection

376

00:17:09,040 --> 00:17:07,010

and icy world and the moon okay it's not

377

00:17:11,110 --> 00:17:09,050

a contradictory okay thanks

378

00:17:14,230 --> 00:17:11,120

I'd like to encourage those with extra

379

00:17:17,170 --> 00:17:14,240

questions to talk to the speaker after

380

00:17:18,589 --> 00:17:17,180

the session so let's thank all the

381

00:17:19,919 --> 00:17:18,599

speakers